

Hiilidioksidin hyötykäytön teknistaloudellinen kannattavuus – Case sellutehdas

BioCO₂-projektin loppuseminaari -
30. elokuuta 2018, Jyväskylä

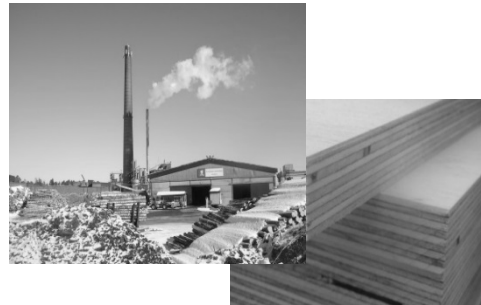
Markus Hurskainen

Case studies

Purpose: study the economic feasibility of CCU processes at different biobased industry sectors

Case 1: Sawmill case

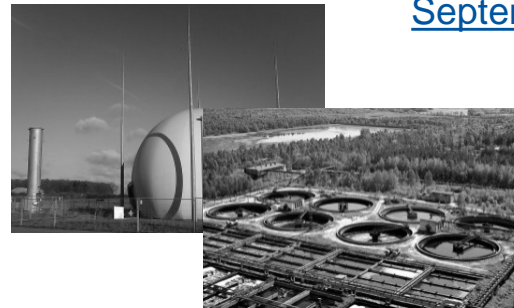
- a) Methane
- b) Methanol
- c) Formic acid



Previous cases presented in the first workshop on September 7th 2017

Case 2: Boosting biogas production

- a) Stand-alone biowaste digester
- b) Waste-water treatment plant



- **Case 3: Pulp mill**

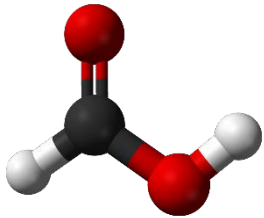
- a) Formic acid
- (b) Methanol)



Kestävää kasvua ja työtä -ohjelma

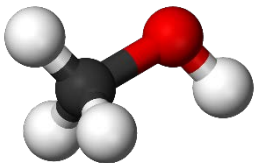
Why these products?

a) Formic acid (HCOOH)



- High-value chemical with low risk of substitution: used e.g. in animal feed, leather tanning, textile dyeing and for producing de-icing agents
- 15% of global (=30% of European) production in Finland
- Theoretically attractive synthesis route from CO₂ and H₂ (no by-products, CO₂+H₂→HCOOH)
- Currently limited market size & low technology readiness level (TRL3-4)

b) Methanol (CH₃OH)



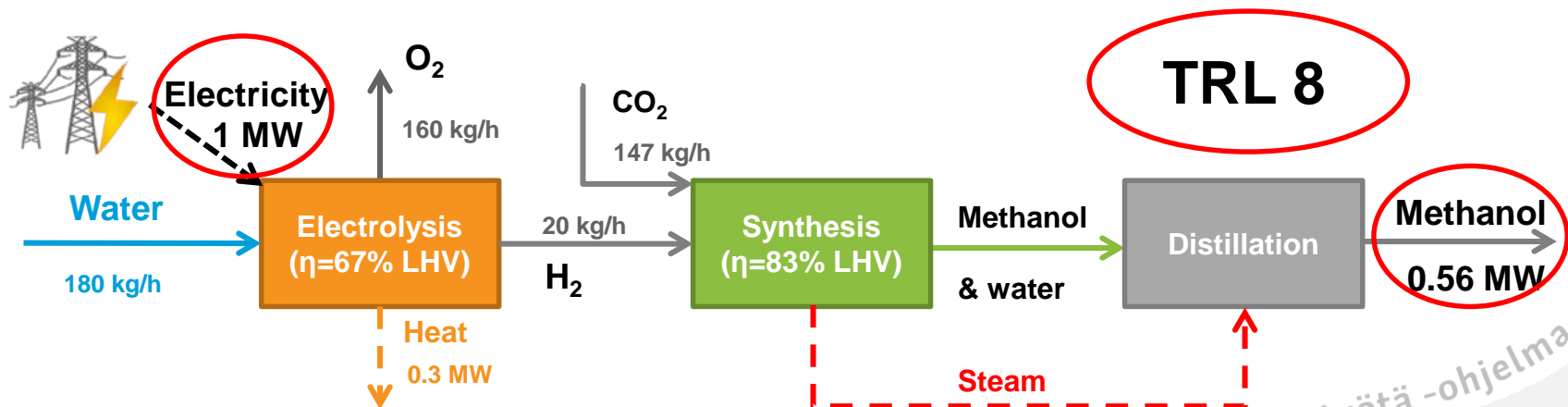
- Large market: important bulk chemical & chemical intermediate, can also be as fuel
- Methanol is used in production of resins/adhesives for wood industry & as a reducing agent in making bleaching chemicals at pulp mills
- Methanol is imported to Finland
- High technology readiness level (TRL8)
–demonstrated at relevant scale already

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Methanol production from CO₂ and H₂

Mass and energy balances (1 MW_e electrolyser)

- Catalytic chemical conversion of carbon dioxide and hydrogen
 - $\text{CO}_2 + 3\text{H}_2 \rightarrow \text{CH}_3\text{OH} + \text{H}_2\text{O}$
 - Copper and zinc based catalysts, at ~250°C and 50-100 bar
- Distillation to separate methanol and water



- Based on Carbon Recycling International/Mitsubishi Hitachi Power Systems Europe process.

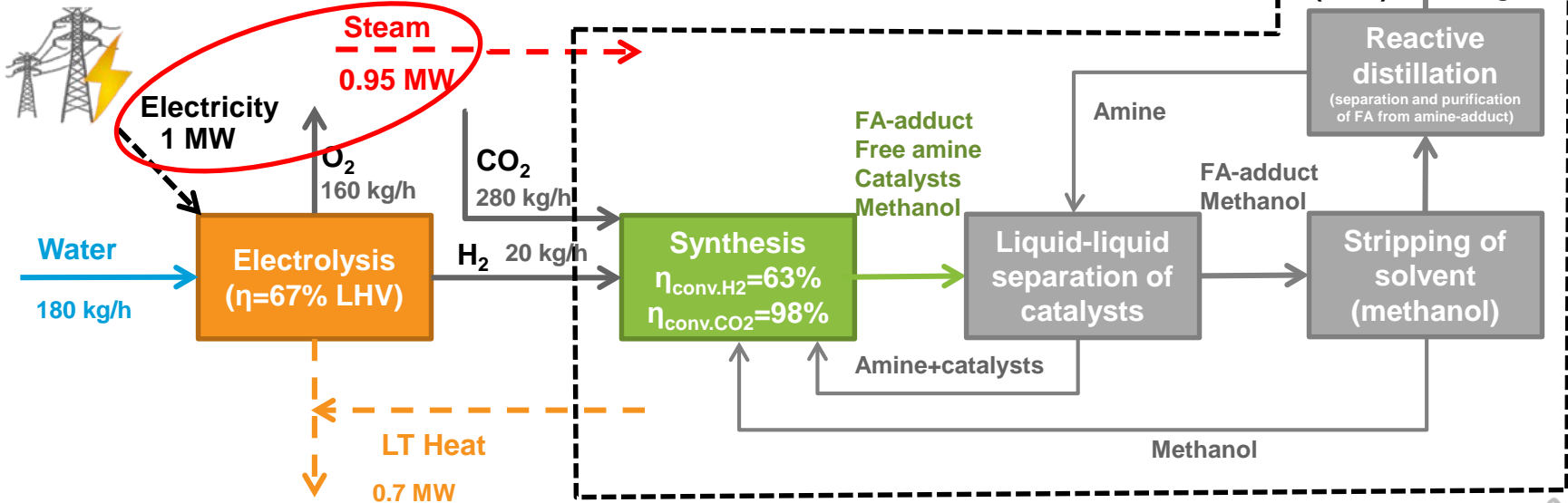
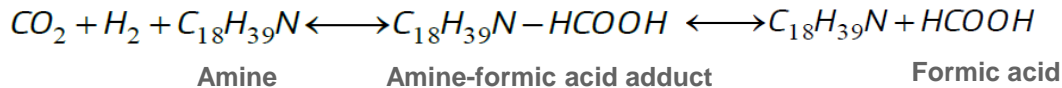
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Formic acid production from CO₂ and H₂

Simplified block diagram (1 MW_e electrolyser)

TRL 3-4

- Chemical conversion of CO₂ and H₂ via homogenous catalysis



- 105 bar & ~90 °C, ruthenium- and phosphino-based catalysts
- Tertiary amine for adduct-formation and polar solvent (methanol-water)
- Steam required for separation of solvent and amine from the product

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Pulp mill

Attractive integration possibilities for formic acid production

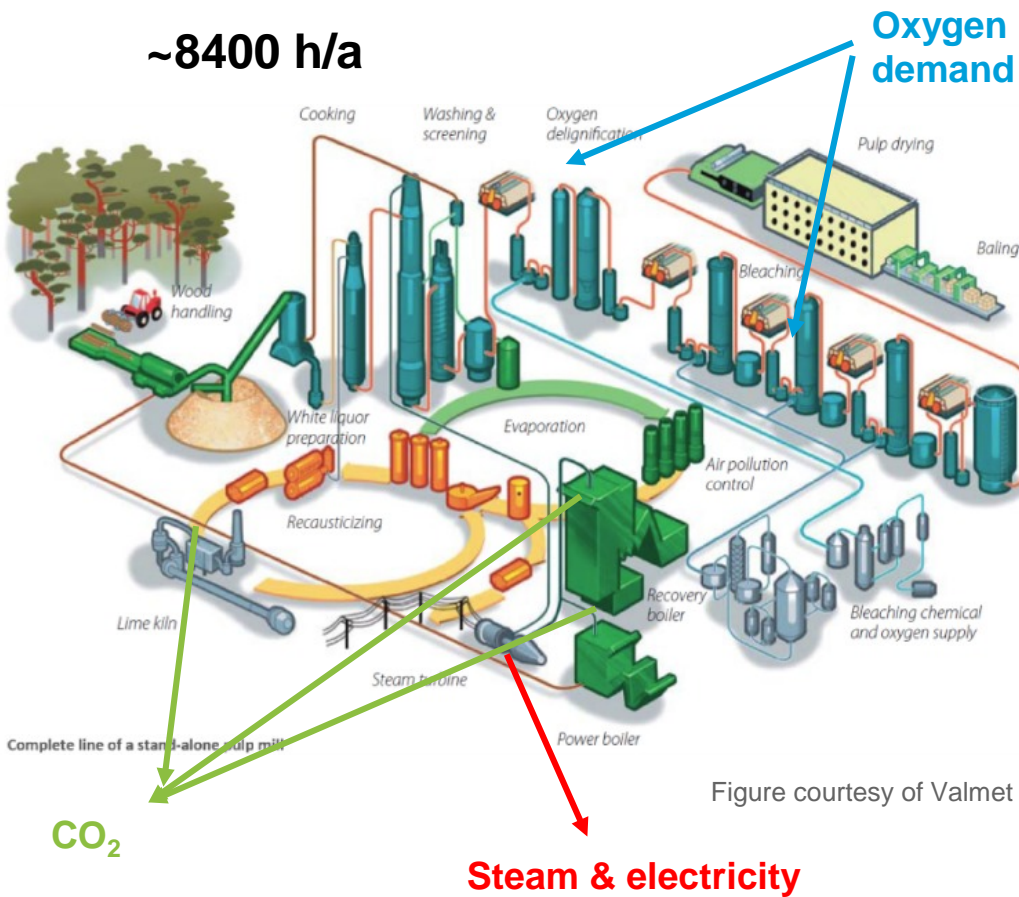


Figure courtesy of Valmet

- Demand for oxygen (delignification & bleaching)
- Low-cost steam available
- Several CO₂ sources and CO₂ could be used also for other purposes (pH control, lignin precipitation, production of PCC* ...)
- As formic acid production is less dependent on the electricity price, high capacity factor of pulp mills can be made use of

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Pulp mill

Reference pulp mill

- Reference **Nordic pulp mill** with annual capacity of 800 000 ADt/a bleached **softwood kraft pulp (BSKP)** as described in IEAGHG (2016) **scaled to Äänekoski capacity (1 300 000 ADt/a)**
- Full load hours ~8400 h/a
 - Forced two-week shutdown in the summer time
- Specific oxygen demand 35 kg/ADt
 - Based on Äänekoski environmental permission



IEAGHG (2016). TECHNO-ECONOMIC EVALUATION OF RETROFITTING CCS IN A MARKET PULP MILL AND AN INTEGRATED PULP AND BOARD MILL

Pulp mill

Formic acid plant scale

	CO ₂ available from pulp mill			CO ₂ demand for formic acid			t/h
	Recovery boiler	Bark boiler	Lime kiln	Electrolyser 10 MW _e	Electrolyser 50 MW _e	Electrolyser 100 MW _e	
Total CO ₂	318	58	42	2.8	14	28	

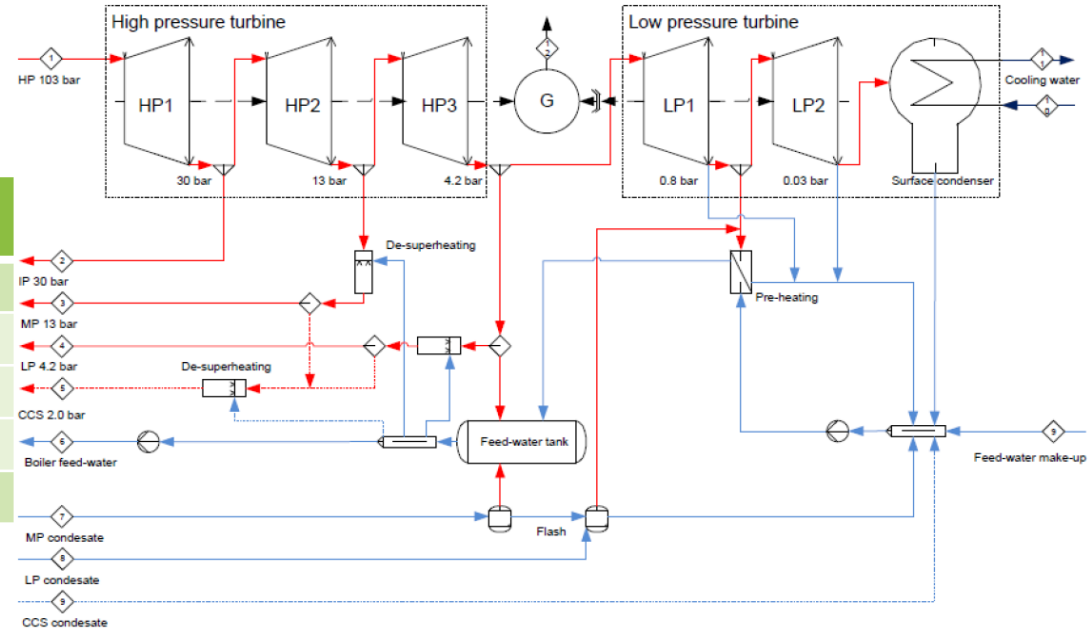
- There is plenty of CO₂ available
 - Capacity for the formic acid plant selected based on the O₂ demand of the mill
 - **32.5 MWe electrolyser (9.2 t_{CO2}/h)**

→ **Formic acid plant capacity ~100 000 t/a** (@ 85% concentration, 8760 h/a)

~75% of Oulu plant capacity

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Pulp mill Electricity and steam



Electricity	Value	Unit
Consumption	61	MW
Gross production	168	MW
-HP section	130	MW
-LP section	38	MW
Net production	107	MW

- Mill could easily provide the needed electricity & process steam (13 & 4.2 bar)

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Pulp mill

Values of by-products and utilities

- Value of **steam** is calculated based on lost electricity production due to process steam bleeding
- There is no use for the **low-temperature heat** → 0 €/MWh
- Value of **oxygen**: Total production costs of oxygen with on-site VPSA oxygen plant (=replacing the old oxygen plant or a greenfield mill)
- Value of **electricity** is Nordpool spot price (2016) + net tax
 - Assumption: no transmission costs (pulp mill provides the electricity)

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Pulp mill

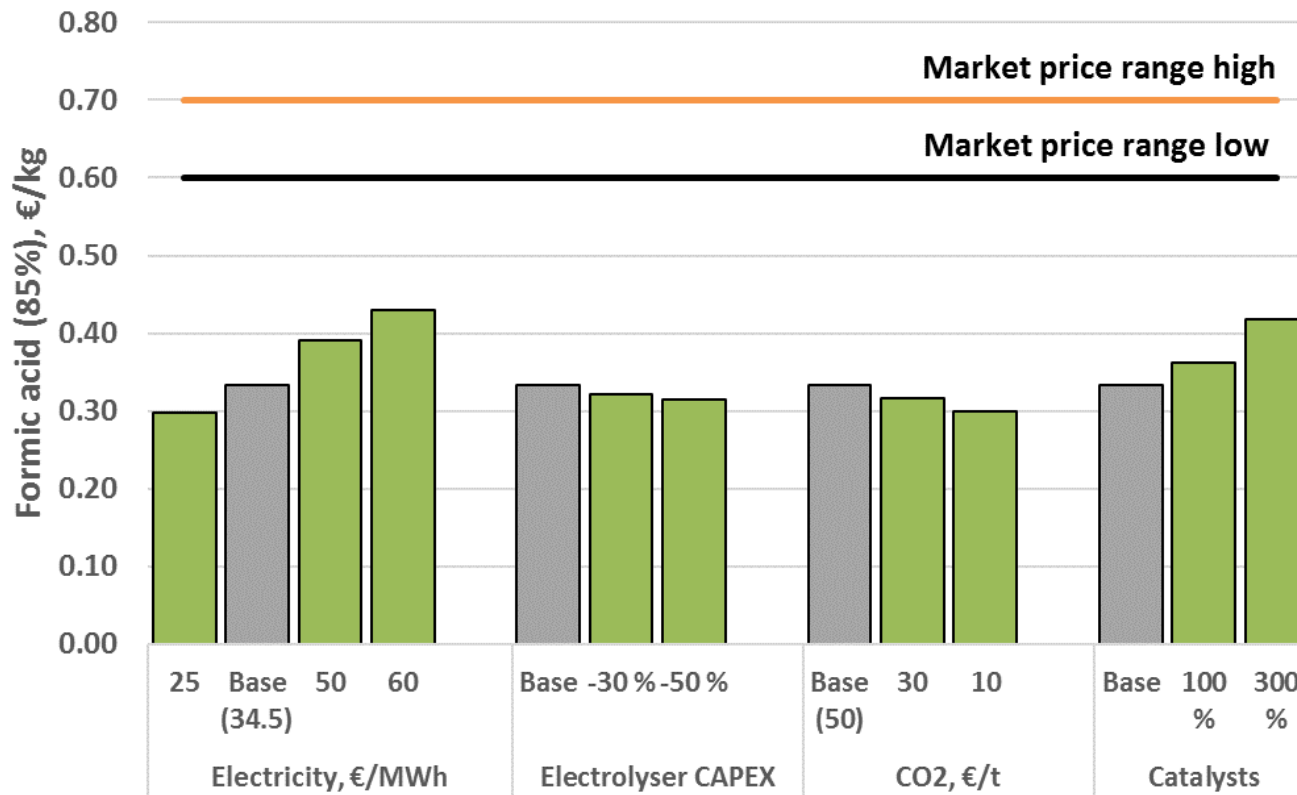
Main market parameters

Parameter	Value
Electricity spot price scenario*	Finland 2016 (avg. price 32.0 €/MWh)
Electricity transmission + net taxes	2 €/MWh (only net tax after 85% tax return)
FCR scenario	-
CO ₂ capture+purification*	50 €/t _{CO2}
O ₂ utilisation	Avg 41 €/t _{O2}
Heat utilisation	0 €/MWh
Cost of steam**	Avg 8.1 €/MWh

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Results

Production costs – Formic acid

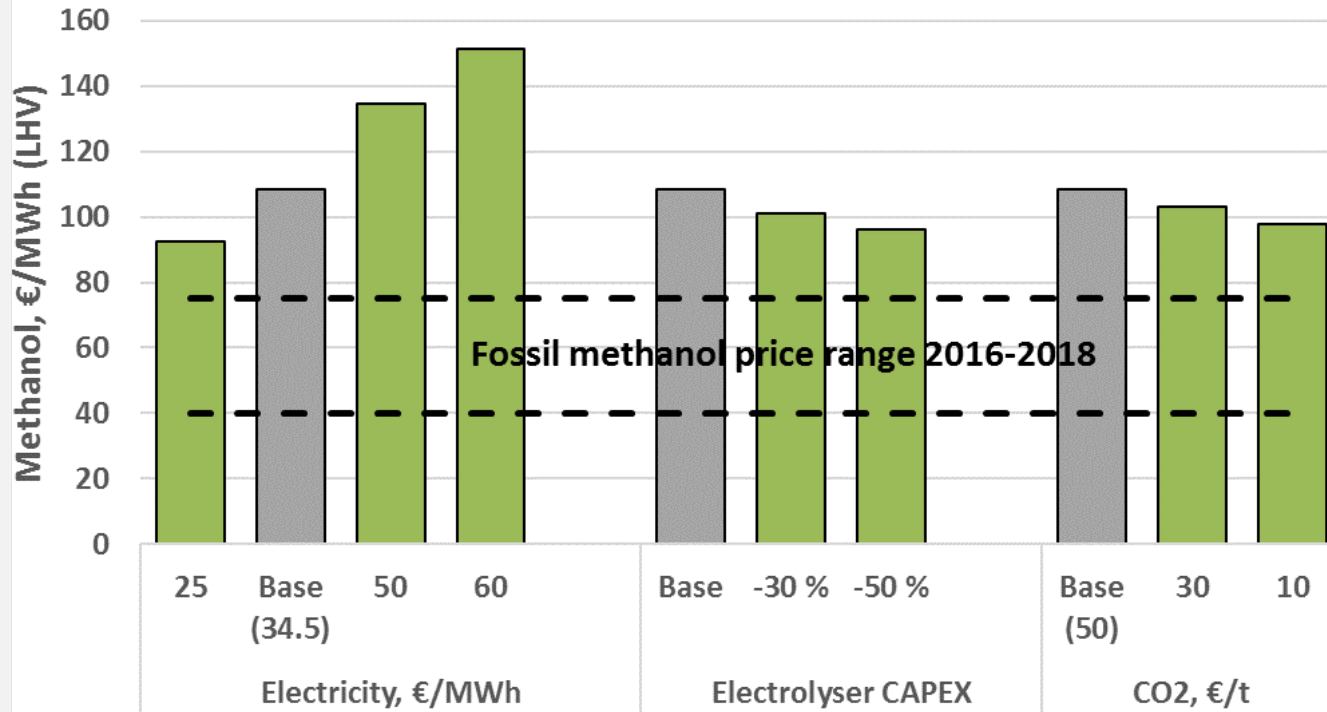


- Production costs only around half of the market prices
- If electricity price would be 50 €/MWh, product value 500 €/t and catalyst costs 100% higher, payback time would be ~7 years (corresponding to 425 €/t production cost)

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Results

Production costs – Methanol



- Production cost ~1.5-2.5 times more than the price of fossil methanol in 2016-2018

→ Required premium for 'green' methanol in the range of 35-70 €/MWh for the base case

- Electricity price is the dominant factor

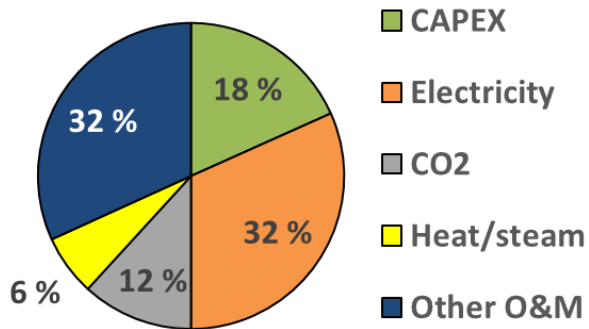
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Results

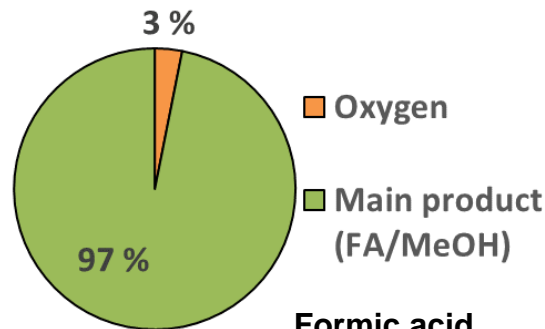
Cost and income breakdowns

Formic acid

Production cost breakdown

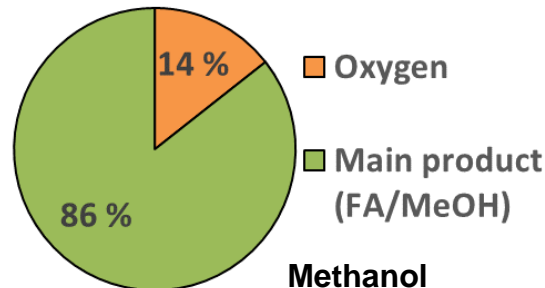
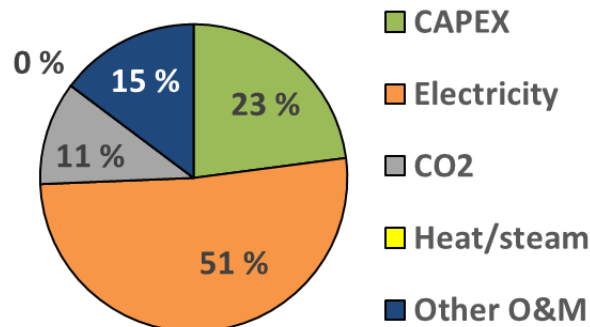


Income breakdown



Formic acid
@ 600 €/t

Methanol



Methanol
@ 70 €/MWh

- Catalyst and amine costs cause high other O&M costs for formic acid production
- Electricity accounts for half of the costs for methanol
- Oxygen income clearly more important for methanol: lower value end-product

... kasvua ja työtä -ohjelma

Conclusions

- CO₂-to-formic acid process integrated to a pulp mill showed promising economics but..
 - Technical feasibility of CO₂ hydrogenation to formic acid still not proven
 - Limited market demand: new formic acid plants are mainly of interest when a current plant is at the end of its lifetime?
- Methanol production would require high 'green premium'

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Kiitos!

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